

Amended Claims

1. A method for electrolytic production of aluminium metal from an electrolyte (3) comprising aluminium oxide, by performing electrolysis comprising at least one inert anode (1) and at least one cathode (2) thus forming part of an electrowinning cell, where the anode evolves oxygen gas and the cathode has aluminium discharged onto it in the electrolysis process, where the said oxygen gas enforces an electrolyte flow pattern,
c h a r a c t e r i s e d i n t h a t
the cathode is substantially horizontal and that oxygen gas is directed to flow into grooves in the electroactive surface of the anode(-s), and drained away from the interpolar room, and thereby establishing an electrolyte flow pattern between the electrodes (1) and (2) and between and over the anode(-s) (1).
2. A method in accordance with claim 1,
c h a r a c t e r i s e d i n t h a t
the anode(-s) and/or the anode connections can be cooled to provide heat exchange and/or heat recovery from said anode/cathode, and/or temperature control.
3. A method in accordance with claim 1,
c h a r a c t e r i s e d i n t h a t
the said anode(-s) and/or the anode connections can be cooled by means of water cooling or other liquid coolants, by gas cooling or by the use of heat pipes.
4. A method in accordance with claim 1,
c h a r a c t e r i s e d i n t h a t
the feeding of alumina to the cell is preferentially continuous or in very small batches (semi-continuous) where the alumina fed to the cell should contain as fine particulates as possible.

5. A method in accordance with claim 1,
characterised in that
the said cell uses an electrolyte with a temperature in the range 880 - 970°C .
6. An electrowinning cell for electrolytic production of aluminium metal from an electrolyte (3) comprising aluminium oxide, the cell comprising at least one inert anode (1) and at least one cathode (2), where the anode evolves oxygen gas and where the cathode has aluminium discharged onto it in the electrolysis process, the said oxygen gas enforces an electrolyte flow pattern,
characterised in that
the cathode is substantially horizontal and that the anode(-s) has grooves in the electroactive surface thereof, where the electrolyte flow pattern is established between the electrodes (1) and (2) and between and over the anode(-s) (1).
7. An electrowinning cell in accordance with in claim 6,
characterised in that
the said anode(-s) are shaped to form "teeth" separated by 1-3 cm deep and 1-3 cm wide grooves.
8. An electrowinning cell in accordance with claim 7,
characterised in that
the bottom of the said anode teeth is V-shaped and sloped 1-5 ° from the centre line towards the groove (4) to efficiently drain the produced gas into the groove.

9. An electrowinning cell in accordance with claim 7,
characterised in that
the surface of the said anode teeth should be horizontally oriented or angled $1-2^{\circ}$,
while the bottom of the grooves in the said anode should be sloped $1-5^{\circ}$ and
oriented parallel to the desired bath circulation pattern to obtain efficient drainage
of produced gas collected in the grooves (4) and to set up a desired flow pattern in
the electrolyte (3).
10. An electrowinning cell in accordance with claim 7,
characterised in that
the said anode "teeth" should be 10-20 cm wide to obtain a uniform current density
and a low bubble layer resistance, where the length of the teeth are not limited, and
can be more than 100 cm.
11. An electrowinning cell in accordance with claim 6,
characterised in that
the corners and edges on the said anode and grooves are smoothened/rounded to
give a uniform flow characteristic and current density.
12. An electrowinning cell in accordance with claim 6,
characterised in that
the top surface of the said anode (13) should be shaped to set up a circulation
pattern that distributes fresh electrolyte to all parts of the cell.
13. An electrowinning cell in accordance with claim 6,
characterised in that
the top of the said anode should be insulated (9) above the bath level around the
stubs as well as the cathode bottom (7) to make it possible to run the cell thermally
in balance with reduced inter polar distance compared to traditional Hall-Heroult
cells.

14. An electrowinning cell in accordance with claim 6,
characterised in that
the anode (1) preferentially should be totally immersed in the electrolyte (3) to
achieve sufficient electrolyte flow and thermal balance in the cell.
15. An electrowinning cell in accordance with claim 6,
characterised in that
two or more anodes form an anode "cluster" which are connected to anode raisers
(6) and via the anode beam to the busbar system in a similar way as prebake
carbon anodes do in Hall-Heroult cells of today.
16. An electrowinning cell in accordance with claim 15,
characterised in that
the said anode clusters are placed with orientation of the grooves in such a way
that the produced oxygen in the grooves sets up an electrolytic flow pattern that
facilitates sufficient electrolytic flow velocity to obtain uniform distribution of
alumina in the cell without muck formation.
17. An electrowinning cell in accordance with claim 15 or 16,
characterised in that
the said anode clusters' position is optimised with respect to groove orientation and
side and centre channels to give the necessary alumina mixing and distribution.
18. An electrowinning cell in accordance with claim 6,
characterised in that
the bottom of the anode facing the cathode can be shaped like a cone or a roof with
3 or more planes with angled or straight surfaces towards a hole (16) in the top
where produced anode gas can escape.

19. An electrowinning cell in accordance with claim 6,
characterised in that
the said anodes are manufactured from dimensionally stable materials, preferably
oxide based cermets, metals, metal alloys, oxide ceramics, and combinations or
composites thereof.
20. An electrowinning cell in accordance with claim 6,
characterised in that
the said anode(-s) can be made of a ceramic outer surface with a good electrical
conducting material made of a cermet or a metal or a combination thereof in the
centre.
21. An electrowinning cell in accordance with claim 6,
characterised in that
the said anode(-s) consisting of a plurality of smaller units integrated in one larger
unit.
22. An electrowinning cell in accordance with claim 6,
characterised in that
the said cell is connected to at least one gas exhaust system for extracting and
collecting gases from the electrolysis chamber.
23. An electrowinning cell in accordance with claim 6,
characterised in that
it comprises the exhaust system which is connected to the alumina feeding system
(11) in which the hot off-gases are used for heating the alumina feed stock and/or
used for scrubbing cleaning of the off-gasses from the cell to remove fluoride
vapours, fluoride particulates and/or dust before entering the gas collection system.

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24. An electrowinning cell in accordance with claim 6,
characterised in that
the cathodes are manufactured from carbon blocks or carbon covered or mixed
with an electrically conductive refractory hard materials (RHM) based on borides,
carbides, nitrides, silicides or mixtures thereof.
25. An electrowinning cell in accordance with claim 6,
characterised in that
the cathode is made of horizontal carbon blocks or a drained carbon composite
blocks.
26. An electrowinning cell in accordance with claim 6,
characterised in that
the aluminium pool acting as a cathode is stabilised by optimising the busbar
system magnetic field.
27. An electrowinning cell in accordance with claim 6,
characterised in that
the cell has a sidewall lining that preferably consists of an electrically non-
conductive material.
28. An electrowinning cell in accordance with claim 6,
characterised in that
the material of the cell sidewall lining is selected from aluminium oxide,
aluminium nitride, silicon carbide, silicon nitride, and combinations thereof or
composites thereof.

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29. An electrowinning cell in accordance with claim 6,
characterised in that
the anodes and/or cathodes are connected to a periphery busbar system for
electrical supply of cells arranged "end-to-end" or "side-by-side".
30. An electrowinning cell in accordance with claim 6,
characterised in that
it comprises at least one feeding point (11) for alumina which is located at a
position being close to high-turbulence areas in the electrolyte, and in the area
between two or more of the said anodes.
31. An electrowinning cell in accordance with claim 6,
characterised in that
the electrolyte comprises a mixture of sodium fluoride and aluminium fluoride,
with possible additional metal fluorides of the group 1 and 2 elements in the
periodic table according to the IUPAC system, and the possible components based
on alkali or alkaline earth halides up to a fluoride/halide molar ratio of 2.5, and
where the NaF/AlF₃ molar ratio is in the range 1 to 3, preferably in the range 1.2 -
2.8.